



# **MAE 598: Multi-Robot Systems**

## **Fall 2016**

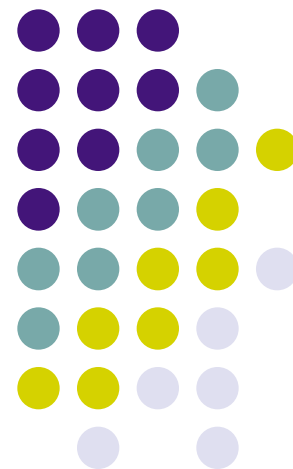
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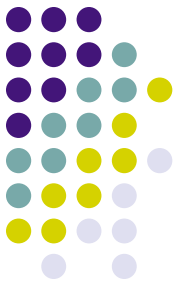
**Lecture 6**

# Classifying Dynamical Behavior of Chemical Reaction Networks

Spring Berman



# Motivation



## ➤ Analysis

Understand cell functions at the level of chemical interactions [Angeli, de Leenheer, Sontag, *CDC 2006*]

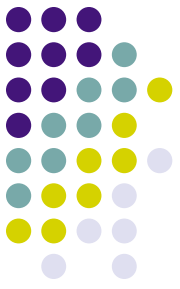
- functionality, qualitative behavior of pathways
- robustness of network to parameter changes

## ➤ Synthesis

Determine whether a network will produce the desired behavior, or at least have the capacity to produce it

- drug design, therapeutic treatments
- bio-inspired distributed robot systems

# Approaches



- There is presently no unified theory of the dynamical behavior of chemical reactions  
[De Leenheer, Angeli, Sontag, *J. Math. Chem.* 41:3, April 2007]
- However, there are results for restricted classes of reaction networks:
  - **Feinberg, Horn, Jackson**  
Fairly general network topology, mass-action kinetics
  - **Angeli, de Leenheer, Sontag**  
Restricted network topology, monotone but otherwise arbitrary kinetics



# Feinberg, Horn, Jackson

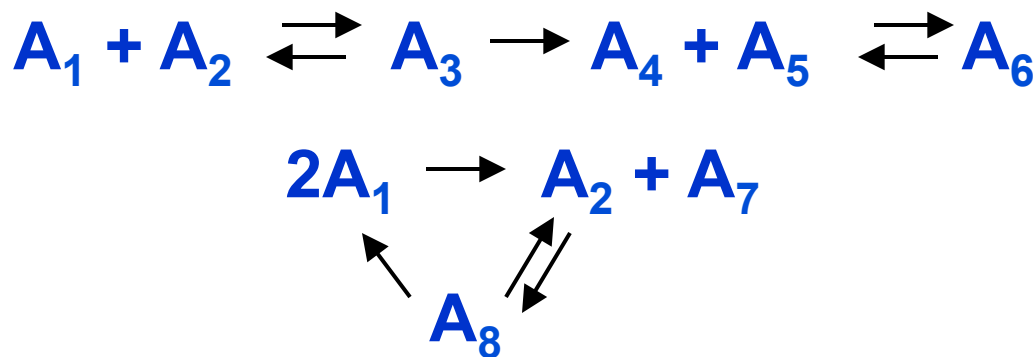
## Deficiency Zero and Deficiency One Theorems

Martin Feinberg. Chemical reaction network structure and the stability of complex isothermal reactors – I. The Deficiency Zero and Deficiency One Theorems. *Chem. Eng. Sci.* 42:10 pp. 2229-2268, 1987.

For related publications, see:

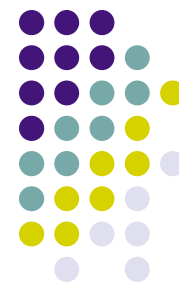
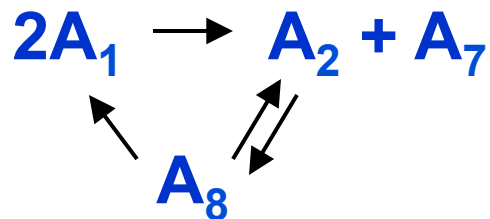
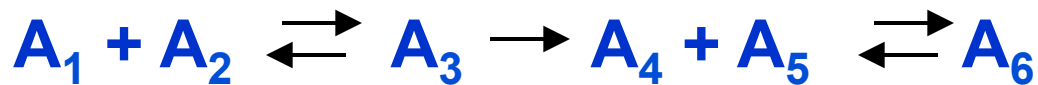
<http://www.che.eng.ohio-state.edu/~FEINBERG/PUBLICATIONS/>

# Notation



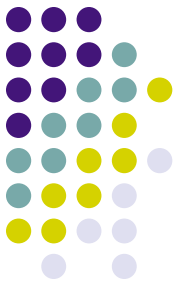
|                           | Symbol                                  | Example above   |
|---------------------------|---|---|
| Number of species         | $N$                                     | <b>8</b>  |
| Number of complexes       | $n$                                     | <b>7</b>  |
| Complex vector            | $y_i \in \mathbb{R}^N$                  | $y_1 = [1 \ 1 \ 0 \ 0 \ 0 \ 0 \ 0 \ 0]$                                   |
| Reaction vector           | For $y_i \rightarrow y_j$ : $y_j - y_i$ | $y_2 - y_1 = [-1 \ -1 \ 1 \ 0 \ 0 \ 0 \ 0 \ 0]$                           |
| Network rank              | $s$                                     | <b>5</b>  |
|                           |   | [ # of elements in largest linearly independent set of reaction vectors ] |
| Number of linkage classes | $l$                                     | <b>2</b>  |
|                           |   | [ set of complexes connected by reactions ]                               |

# Notation

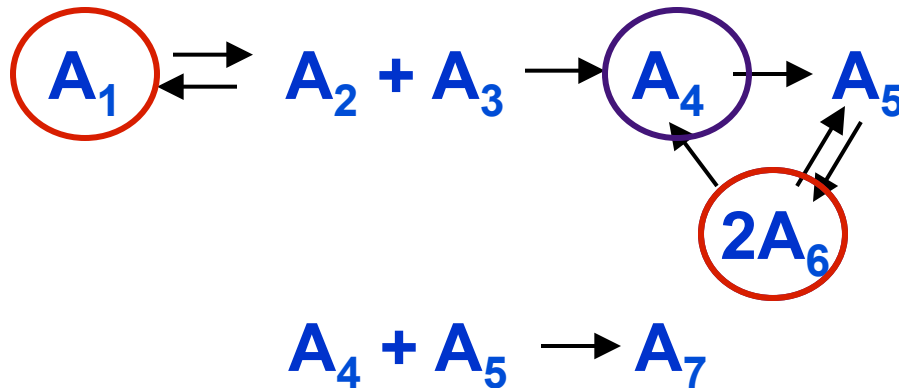


|   | Symbol               | Example above |
|---|----------------------|---------------|
| Number of complexes   | $n$                  | <b>7</b>      |
| Network rank  | $s$                  | <b>5</b>      |
| [ # of elements in largest linearly independent set of reaction vectors ] |                      |               |
| Number of linkage classes   | $l$                  | <b>2</b>      |
| [ set of complexes connected by reactions ]                               |                      |               |
| <b>Deficiency</b>   | $\delta = n - l - s$ | <b>0</b>      |

# Definitions

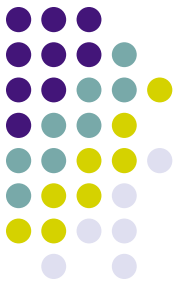


- **Reversible**: Each reaction is accompanied by its reverse
- **Weakly reversible**: When there is a directed arrow pathway from complex 1 to 2, there is one from 2 to 1
- Complexes 1 and 2 are **strongly linked** if there are directed arrow pathways from 1 to 2 and from 2 to 1

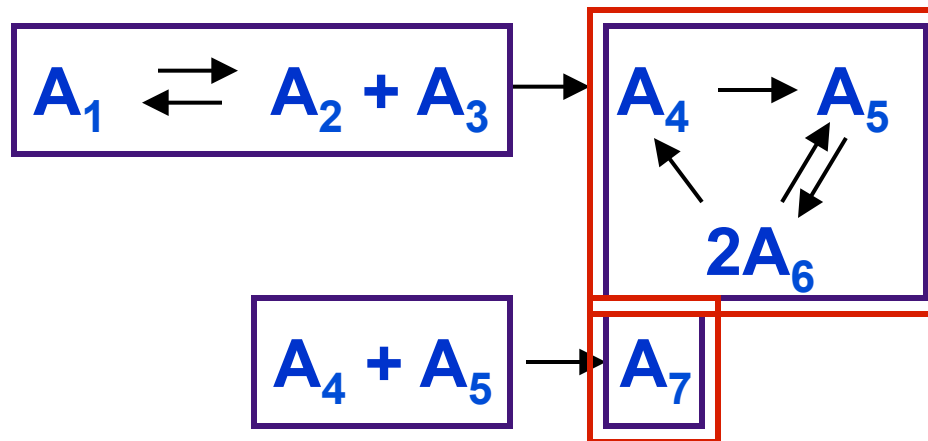




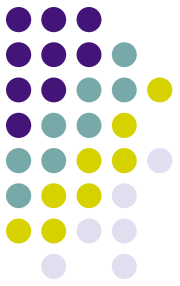
# Definitions



- **Strong linkage class** is a set of complexes for which:
  - Each pair of complexes is strongly linked
  - No complex is strongly linked to a complex outside the set
- **Terminal strong linkage class**: has no complex that reacts to a complex in a different strong linkage class (number =  $L$ )

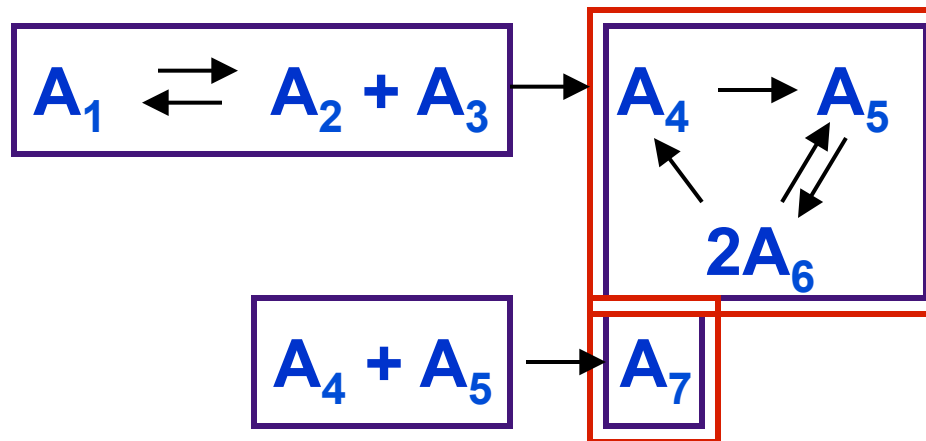


# Remarks

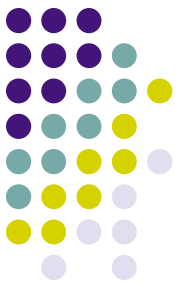


- In general,  $L \geq l$
- For a weakly reversible network,  $L = l$

(Linkage classes, strong linkages classes, terminal strong linkage classes coincide)



# Kinetics, ODE Description



- **Closed**, well-stirred, constant-volume, isothermal reactor
  - Can extend to **open** reactors by adding “pseudoreactions,”  
 $0 \rightarrow A_i, A_i \rightarrow 0$

Species:  $\{A_1, A_2, \dots, A_N\}$

Molar concentration of  $A_i$ :  $c_i \in R_{\geq 0}$

Composition vector:  $c = [c_1 \ c_2 \ \dots \ c_N]$

$P^N$  = positive orthant of  $R^N$        $\underline{P}^N$  = nonnegative orthant of  $R^N$

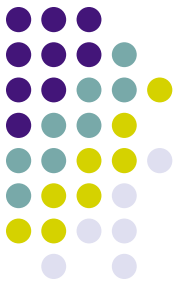
Support of composition vector:  $\text{supp } c = \{A_i \mid c_i > 0\}$

Support of complex:  $\text{supp } y_i = \{A_j \mid y_{ij} > 0\}$



Stoichiometric coefficient

# Kinetics, ODE Description



- Closed, well-stirred, constant-volume, isothermal reactor

Molar concentration of  $A_i$ :  $c_i \in \mathbb{R}_{\geq 0}$

Composition vector:  $\mathbf{c} = [c_1 \ c_2 \ \dots \ c_N]$

- **Kinetics:** An assignment to each reaction  $y_i \rightarrow y_j$  of a rate function  $\mathcal{X}_{i \rightarrow j}(\mathbf{c})$

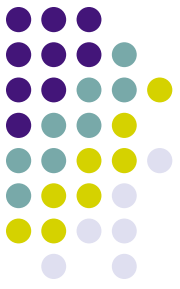
- **Mass action kinetics:** For each reaction  $y_i \rightarrow y_j$ ,

$$\mathcal{X}_{i \rightarrow j}(\mathbf{c}) \equiv k_{i \rightarrow j} \prod_{L=1}^N (c_L)^{y_{iL}}$$

- **ODE Formulation:**

$$\dot{\mathbf{c}} = \sum_{\mathcal{R}} \mathcal{X}_{i \rightarrow j}(\mathbf{c}) (\mathbf{y}_j - \mathbf{y}_i), \quad \mathbf{c} \in \bar{\mathbb{P}}^N$$

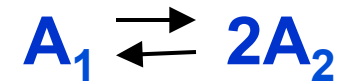
# Properties of ODE's



- Stoichiometric subspace  $S = \{ \boldsymbol{y} \in \mathbb{R}^N \}$ :

$$\boldsymbol{y} = \sum_{\mathcal{R}} \alpha_{i \rightarrow j} (\boldsymbol{y}_j - \boldsymbol{y}_i). \quad \{ \alpha_{i \rightarrow j} \}_{i \rightarrow j \in \mathcal{R}} \geq 0$$

- Network rank  $s = \dim(S)$

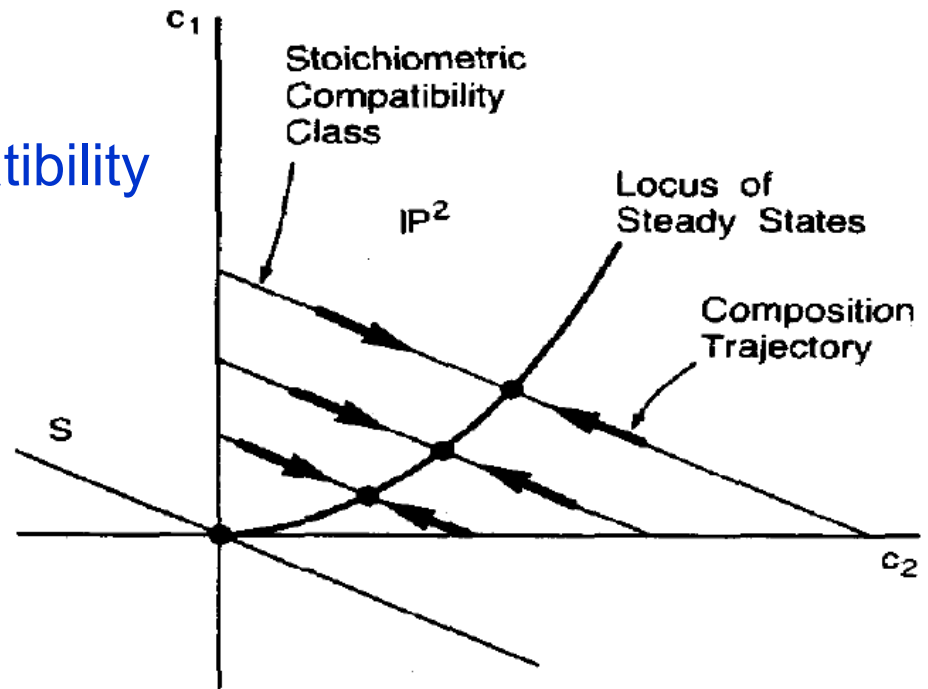


- $\boldsymbol{c}(t) - \boldsymbol{c}(0)$  lies in  $S$

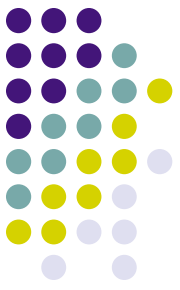
- Positive stoichiometric compatibility class (reaction simplex):

$$(\boldsymbol{c} + S) \cap \mathbb{P}^N.$$

- Goal is to classify dynamics within a stoichiometric comp. class



# Steady States



- Reaction vectors are **positively dependent** if:

$$\mathbf{0} = \sum_{\mathcal{R}} \alpha_{i \rightarrow j} (\mathbf{y}_j - \mathbf{y}_i). \quad \{\alpha_{i \rightarrow j}\}_{i \rightarrow j \in \mathcal{R}} > 0$$

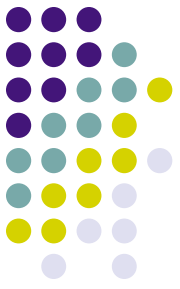
- Always the case in a weakly reversible network

This is a necessary condition for the existence of:

- A **positive steady state**  $\mathbf{c}^* \in \mathbb{P}^N$   $\mathbf{0} = \sum_{\mathcal{R}} \mathcal{K}_{i \rightarrow j}(\mathbf{c}^*) (\mathbf{y}_j - \mathbf{y}_i)$ .
- A **cyclic trajectory**  $\mathbf{c}(\tau) \in \mathbb{P}^N$   $\mathbf{c}(0) = \mathbf{c}(T)$ .

- At steady state, all reactions among complexes in a strong linkage class are switched **on** or **off**

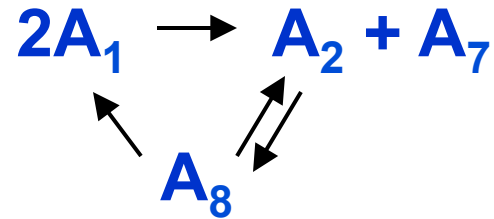
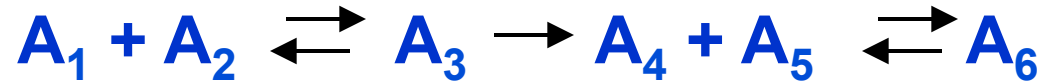
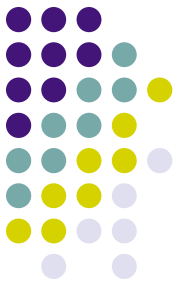
# Deficiency Zero Theorem



When  $\delta = 0$ :

- Network is **not weakly reversible**  
**Arbitrary** kinetics  
→ **No** positive steady state or cyclic trajectory  $\mathbf{c}(\tau) \in \mathbb{P}^N$
- Network **is weakly reversible**  
**Mass action** kinetics  
→ Each positive stoichiometric compatibility class has **one steady state**, which is **asymptotically stable**;  
There is no nontrivial cyclic trajectory  $\mathbf{c}(\tau) \in \mathbb{P}^N$
- *Remark:* The only reactions occurring at steady state are those joining complexes in a terminal strong linkage class

# Deficiency Zero Theorem: Example

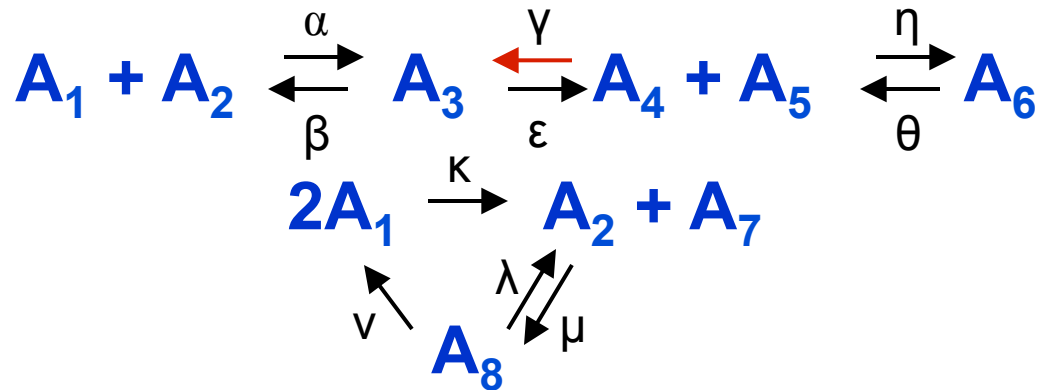


$\delta = 0$ , not weakly reversible

→ **No** positive steady state or cyclic trajectory  $\mathbf{c}(\tau) \in \mathbb{P}^N$



# Deficiency Zero Theorem: Example



$$\dot{c}_1 = -\alpha c_1 c_2 + \beta c_3 - 2\kappa c_1^2 + \nu c_8$$

$$\dot{c}_2 = -\alpha c_1 c_2 + \beta c_3 + \kappa c_1^2 - \lambda c_2 c_7 + \mu c_8$$

$$\dot{c}_3 = \alpha c_1 c_2 + \varepsilon c_4 c_5 - (\gamma + \beta) c_3$$

$$\dot{c}_4 = \gamma c_3 + \theta c_6 - (\varepsilon + \eta) c_4 c_5$$

$$\dot{c}_5 = \gamma c_3 + \theta c_6 - (\varepsilon + \eta) c_4 c_5$$

$$\dot{c}_6 = \eta c_4 c_5 - \theta c_6$$

$$\dot{c}_7 = \kappa c_1^2 + \mu c_8 - \lambda c_2 c_7$$

$$\dot{c}_8 = \lambda c_2 c_7 - (\mu + \nu) c_8$$

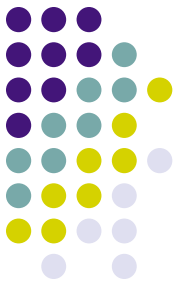
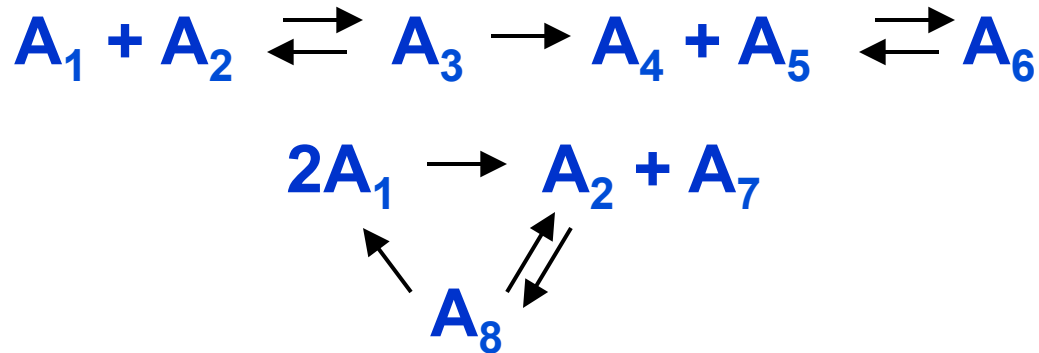
➤ Two networks with the same complexes and linkage classes have the **same deficiency**

$$\rightarrow \delta = 0$$

- Weakly reversible, assume mass action kinetics

→ System has **one positive steady state**, which is asymptotically stable

# Remarks



Deficiency

$$\delta = n - l - s$$

- Two networks with the same complexes and linkage classes have the same **rank** → same **deficiency**
- Network rank  $\leq$  sum of linkage class ranks
- Network deficiency  $\geq$  sum of linkage class deficiencies

# Deficiency One Theorem



Mass action kinetics

$\ell$  linkage classes, each containing one terminal strong linkage class

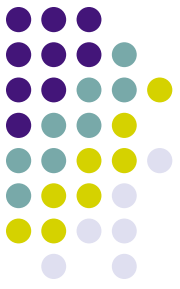
(i)  $\delta_\theta \leq 1, \quad \theta = 1, 2, \dots, \ell$       Linkage class deficiencies

(ii)  $\sum_{\theta=1}^{\ell} \delta_\theta = \delta.$       Network deficiency

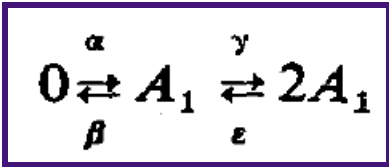
→ **No more than one** steady state in a positive stoichiometric compatibility class (may depend on rate constants)

➤ Network is **weakly reversible**:

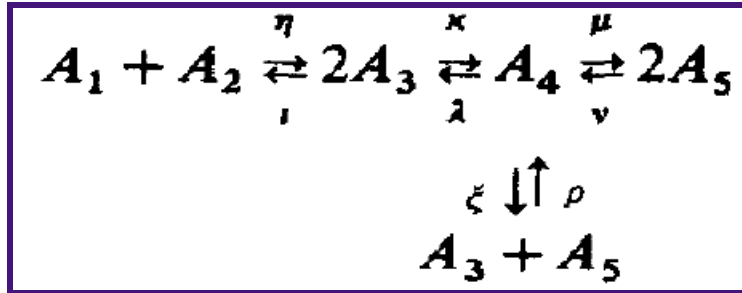
→ **Precisely one** steady state in each pos. stoich. comp. class



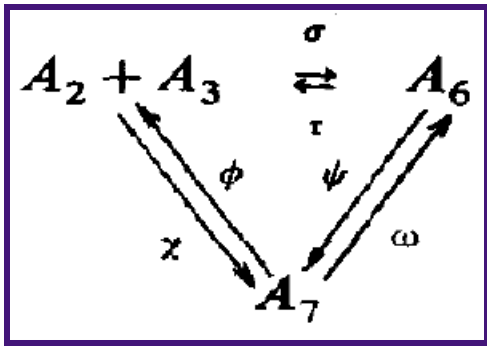
# Deficiency One Theorem: Example



$$\delta_1 = 1$$



$$\delta_2 = 1$$



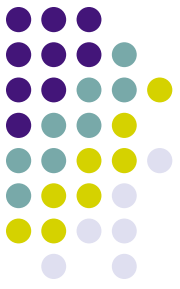
$$\delta_3 = 0$$

$$\delta = 2 = \sum \delta_i$$

➤ Network is weakly reversible

→ Precisely **one** steady state in each pos. stoich. comp. class

# Deficiency One Theorem: Corollary



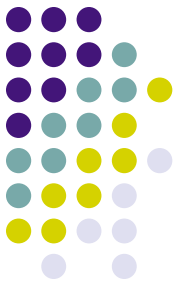
Mass action kinetics

*One* linkage class

$\delta > 1$  or # of terminal strong linkage classes  $L > 1$

→ Can have **multiple steady states** in a pos. stoich. comp. class

# Deficiency One Theorem: Subnetworks



- If a set of reactions is partitioned into  $p$  **subnetworks**, then each is **independent** iff:

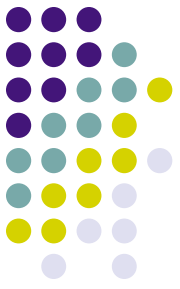
$$s = s_1 + s_2 + \dots + s_p.$$

- A steady state  $\mathbf{c}^*$  for a reaction network is a steady state for any independent subnetwork.

→ Can “carry down” or “carry up” information from Def. Theorems

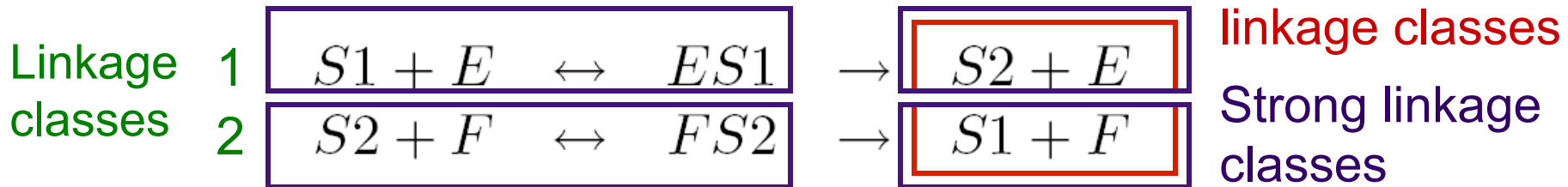
Ex.) Network admits a positive steady state → this is a positive steady state of an independent subnetwork

# Example: Single Phosphorylation



➤ “Futile cycle”

ex) Signaling transduction cascades, bacterial two-component systems



$S1$  = substrate     $S2$  = product     $E, F$  = enzymes     $ES1$  =  $E$  bound to  $S1$

➤ **Not** weakly reversible

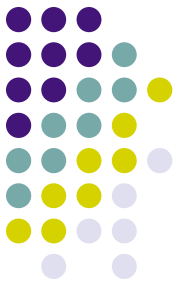
$$\delta = n - l - s = 6 - 2 - 3 = 1 \rightarrow \text{Can't apply Deficiency Zero Theorem}$$

$$\delta_1 = n_1 - 1 - s_1 = 3 - 1 - 2 = 0$$

$$\delta_2 = n_2 - 1 - s_2 = 3 - 1 - 2 = 0$$

$$\delta_1 + \delta_2 \neq \delta \rightarrow \text{Can't apply Deficiency One Theorem}$$

# Deficiency One Theorem: Remarks



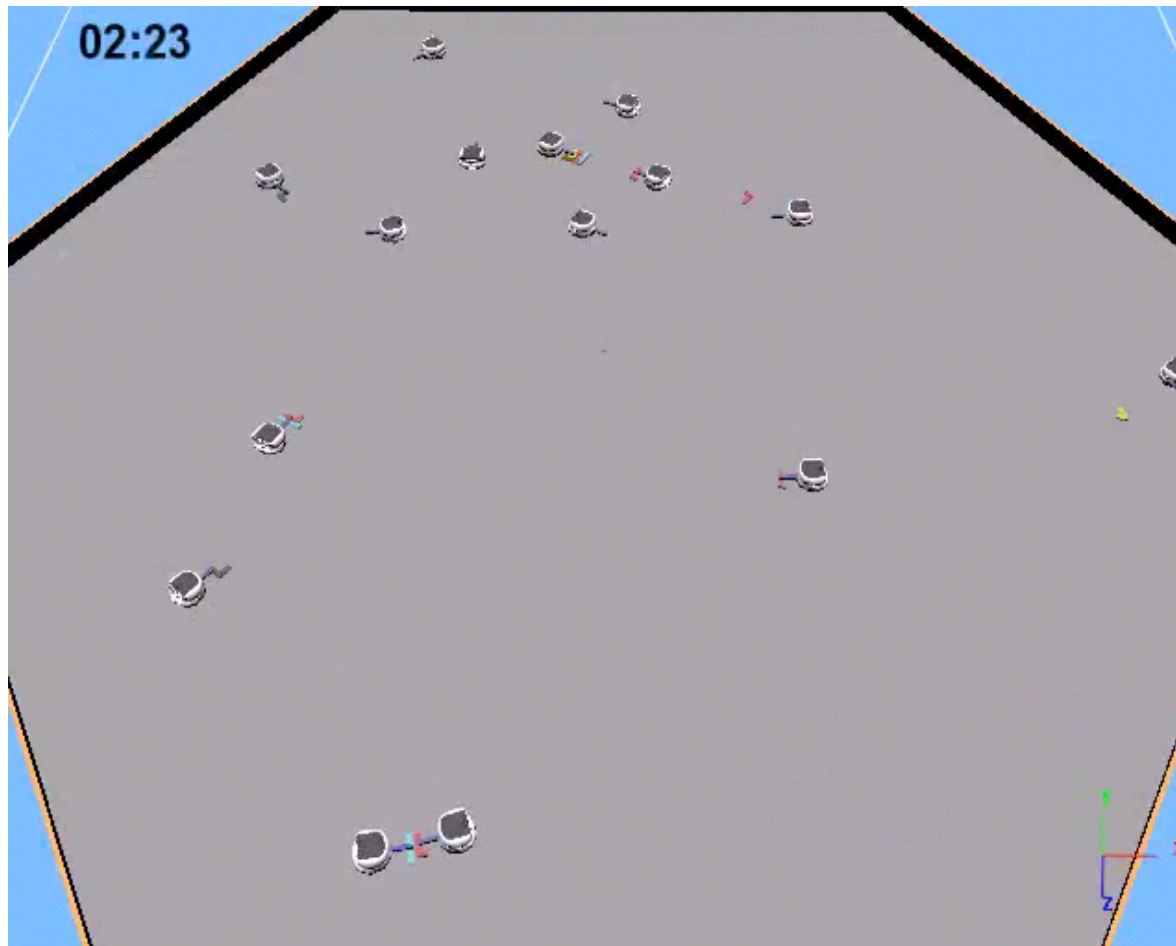
- Deficiency one networks that are not weakly reversible:
  - Can admit positive steady states for some values of **rate constants** *but not for others*
  - Can admit steady states in some **pos. stoich. comp. classes** *but not in others*



# Swarm Robotic Assembly System

*[Matthey, Berman, Kumar, ICRA 2009]*

Design a **reconfigurable manufacturing system** that quickly assembles target amounts of products from a supply of heterogeneous parts



# Required Robot Controller Properties

## (1) Strategy should be **scalable** in the number of parts

### Decentralized decision-making:

- Parts scattered randomly inside an arena
- Randomly moving autonomous robots assemble products
- Local sensing, local communication

## (2) **Minimal adjustments** when product demand changes

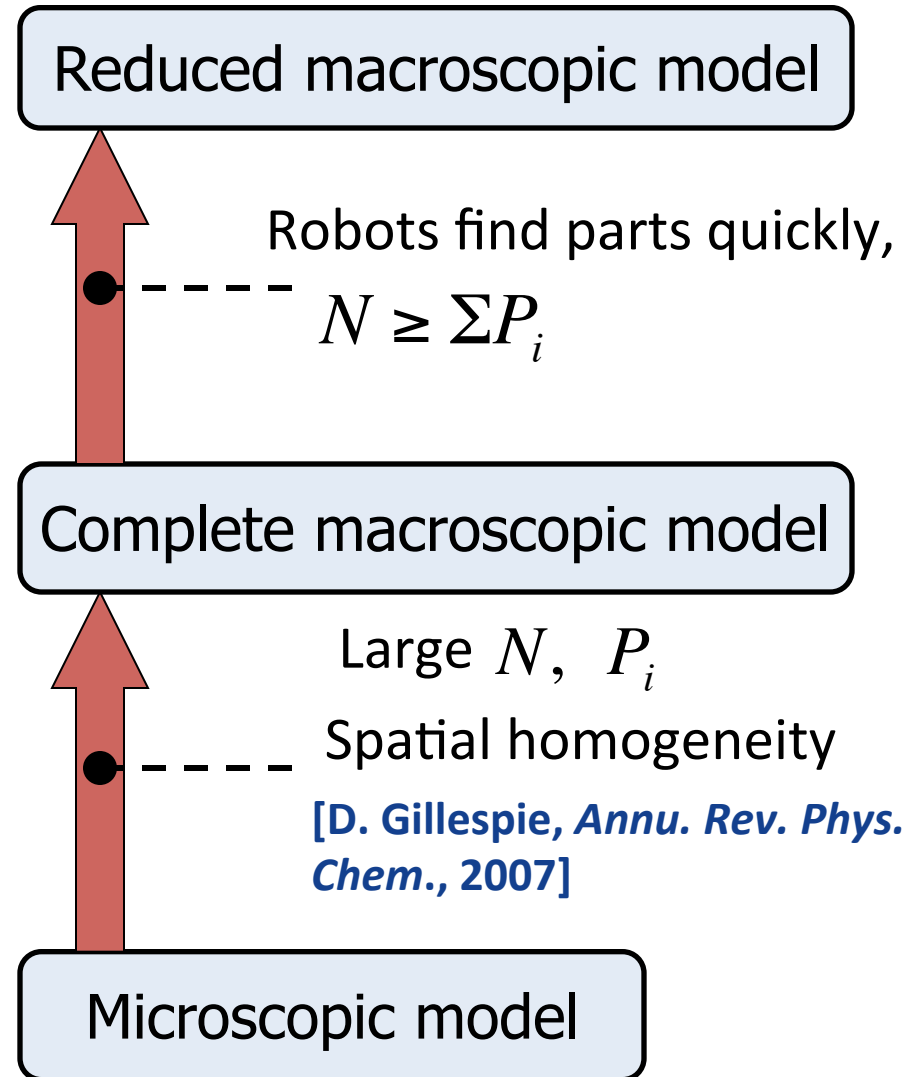
- **Probabilities** of assembly and disassembly are robot control policies
- Can be updated via a broadcast

## (3) System can be **optimized for fast production**

**Spatial homogeneity** → Chemical Reaction Network model

# Approach

- *Ordinary differential equations*  
 **$M$  states**: continuous populations of parts
- *Ordinary differential equations*  
**States**: continuous populations of robots and free/carried parts
- *3D physics simulation*  
 **$N$  robots,  $P_i$  parts;**  
 **$i = 1, \dots, M$  types**



# Approach

ODEs are functions of probabilities of assembly and disassembly:

Optimize for **fast assembly of target amounts of products**

Robots start assemblies and perform disassemblies according to optimized probabilities

```
graph TD; A[Reduced macroscopic model] --> B[Microscopic model];
```

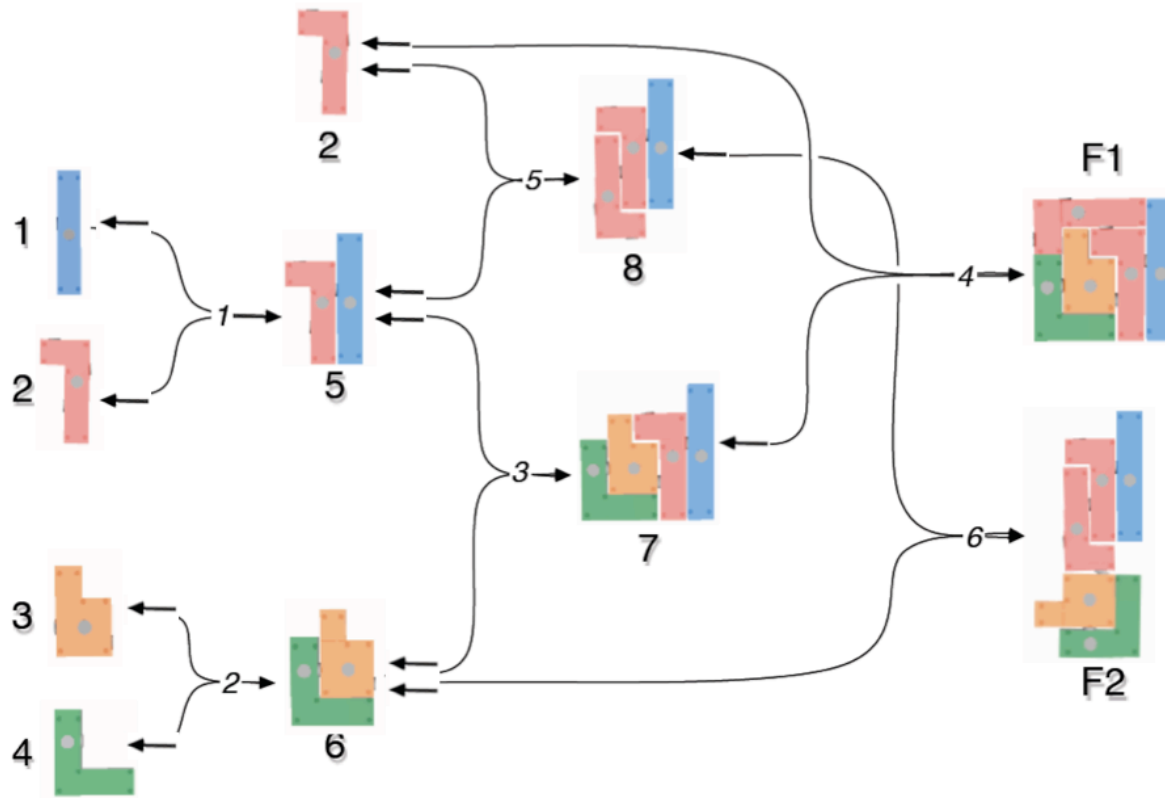
Reduced macroscopic model

Microscopic model

# Example

- Implemented in the robot simulator Webots ([www.cyberbotics.com](http://www.cyberbotics.com))
  - Uses Open Dynamics Engine to simulate physics
- Predefined assembly plan:

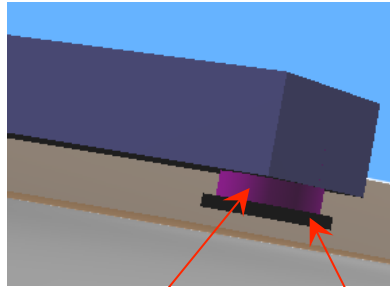
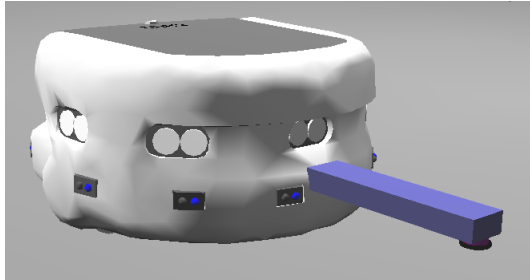
4 types  
of basic  
parts



2 types  
of final  
assemblies

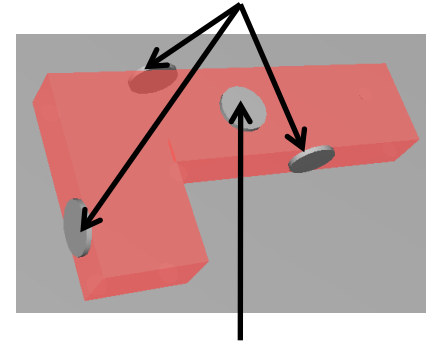
# Example

Khepera III + bar  
([www.k-team.com](http://www.k-team.com))



Rotational Magnet  
servo

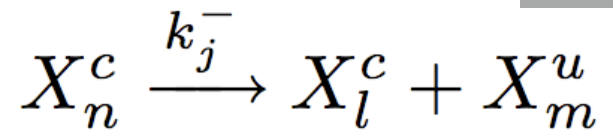
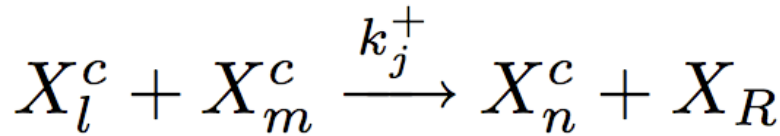
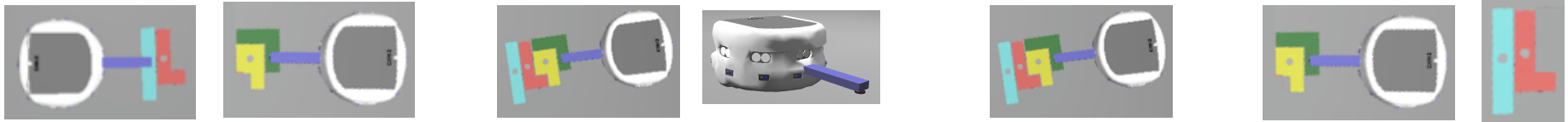
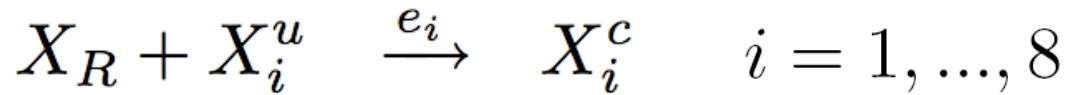
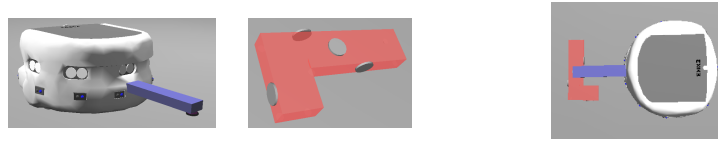
Magnets that bond to  
other parts



Bonds to bar

- Magnets can be turned on or off
- Servo rotates bonded part to orientation for assembly
- Infra-red distance sensors for collision avoidance
- Emitter/receiver on each robot and basic part for local communication, computing relative bearing

# Decisions Modeled as Chemical Reactions

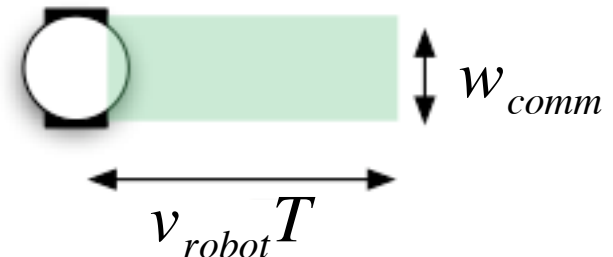


$$e_i = A(p^e), \quad k_j^+ = A(p^e) p_j^a p_j^+, \quad k_j^- = p_j^-$$

$p^e$  = prob. that a robot encounters a part or another robot  $\approx$

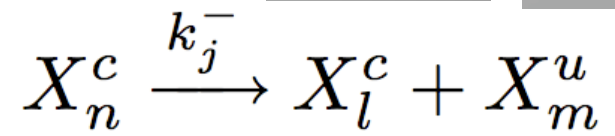
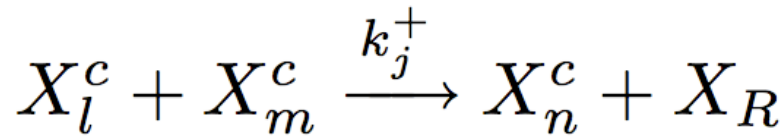
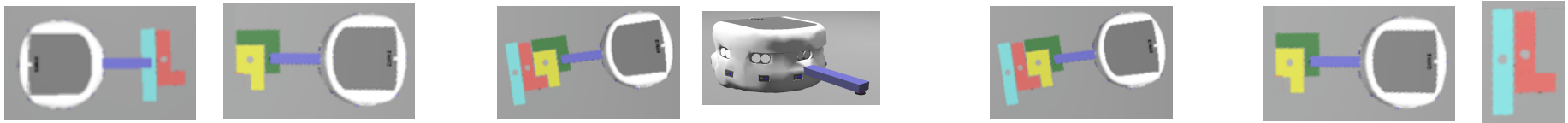
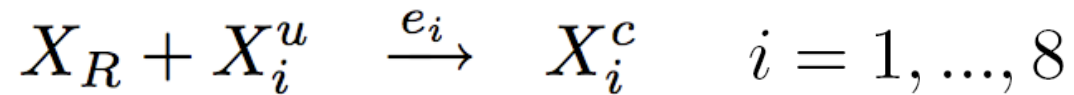
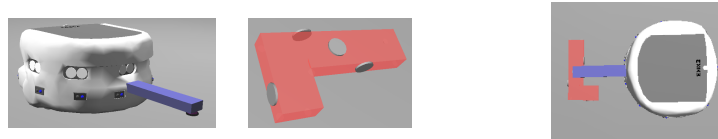
$$\frac{v_{robot} T w_{comm}}{A}$$

$A$  = arena area



[Correll and Martinoli, Coll. Beh. Workshop, ICRA 2007]

# Decisions Modeled as Chemical Reactions

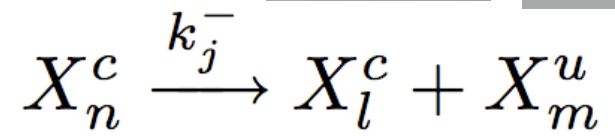
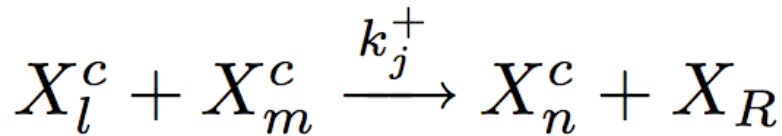
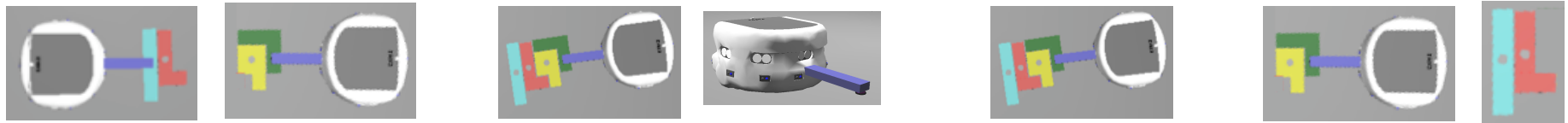
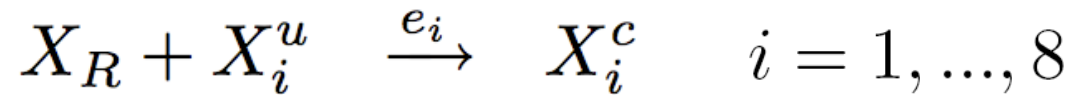
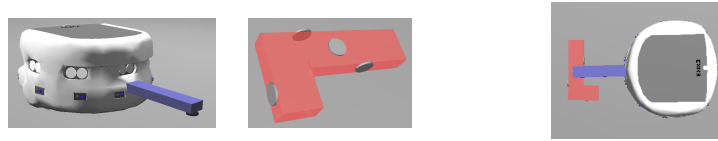


$$e_i = A p^e, \quad k_j^+ = A p^e (p_j^a) p_j^+, \quad k_j^- = p_j^-$$

$p_j^a$  = prob. of two robots successfully  
 completing assembly process  $j$   
 (measured from simulations)



# Decisions Modeled as Chemical Reactions



$$e_i = A p^e, \quad k_j^+ = A p^e p_j^a (p_j^+), \quad k_j^- = (p_j^-)$$

**Tunable:**

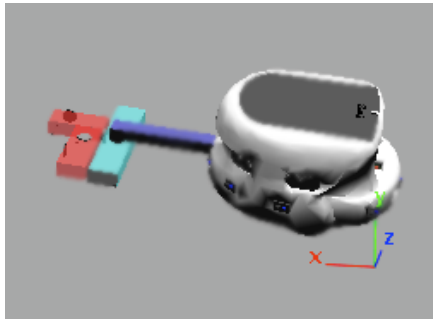
$p_j^+$  = prob. of two robots starting assembly process  $j$

$p_j^-$  = prob. per unit time of a robot performing disassembly process  $j$

# Mapping $p_i^+, p_i^-$ onto the Robot Controllers

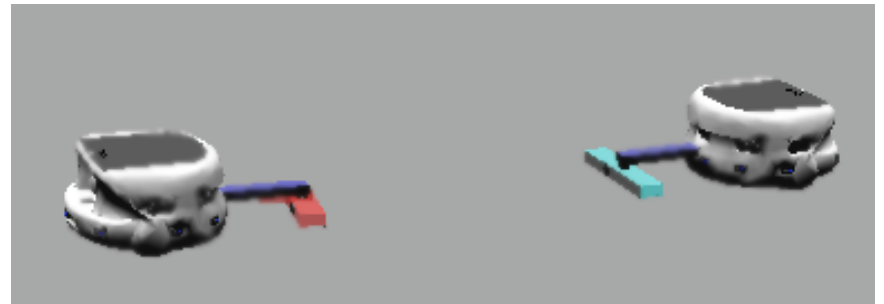
$\Delta t$  = simulation timestep (32 ms)

$u$  = random number uniformly distributed over [0,1]



Robot computes  $u$  at each  $\Delta t$ ,  
disassembles the part if

$$u < p_i^- \Delta t$$

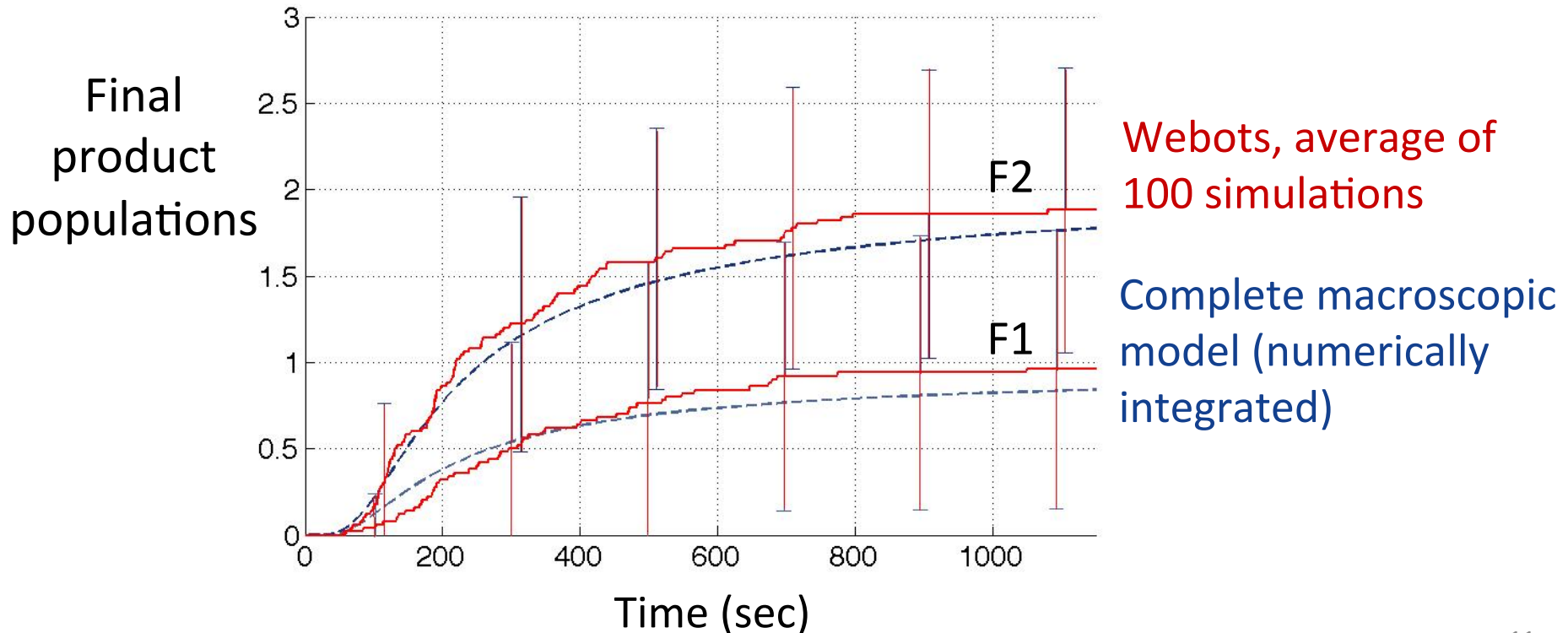


Robot computes  $u$ ,  
executes assembly if

$$u < p_i^+$$

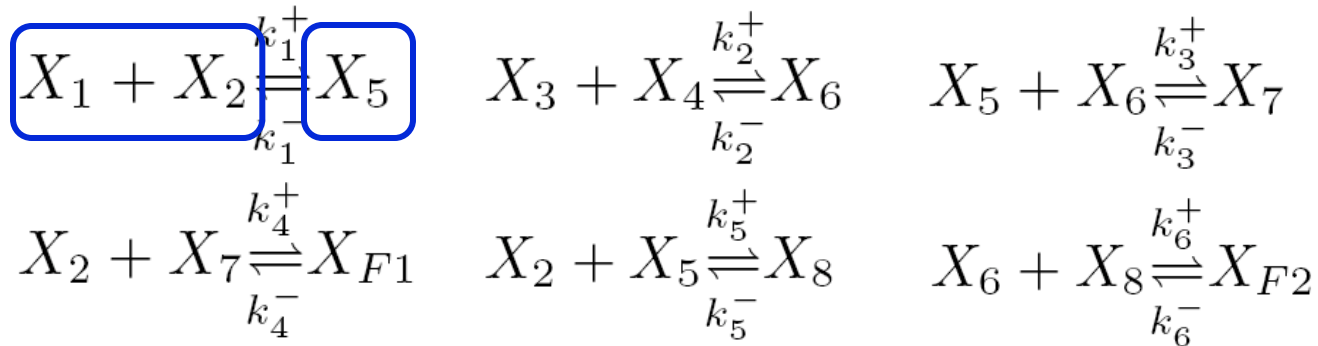
# Validation of Complete Macroscopic Model

- Macroscopic model (set of ODEs) is fairly accurate
- Discrepancies are due to:
  - Relatively low populations; ODE most accurate for large ones
  - Assembly disruption in simulation (not modeled)



# Reduced Macroscopic Model

Lower-dimensional model (abstract away robots):



Vector of complexes:  $\mathbf{y}(\mathbf{x}) = [x_1x_2 \quad x_5 \quad x_3x_4 \quad x_6 \quad x_2x_7 \quad x_{F1} \quad x_5x_6 \quad x_7 \quad x_2x_5 \quad x_8 \quad x_6x_8 \quad x_{F2}]^T$

$$\dot{\mathbf{x}} = \mathbf{MK}\mathbf{y}(\mathbf{x})$$

We also define a matrix  $\mathbf{M} \in \mathbb{R}^{10 \times 12}$  in which each entry  $M_{ji}$ ,  $j = 1, \dots, 10$ , of column  $\mathbf{m}_i$  is the coefficient of part type  $j$  in complex  $i$  (0 if absent). We relabel the rate associated with reaction  $(i, j) \in \mathcal{E}$  as  $k_{ij}$  and define a matrix  $\mathbf{K} \in \mathbb{R}^{12 \times 12}$  with entries

$$\mathbf{K}_{ij} = \begin{cases} k_{ji} & \text{if } i \neq j, (j, i) \in \mathcal{E}, \\ 0 & \text{if } i \neq j, (j, i) \notin \mathcal{E}, \\ -\sum_{(i,l) \in \mathcal{E}} k_{il} & \text{if } i = j. \end{cases} \quad (5)$$

Conservation constraints:

$$\begin{array}{rcl}
 x_3 - x_4 & = & N_1 \\
 x_1 + x_5 + x_7 + x_8 + x_{F1} + x_{F2} & = & N_2 \\
 x_2 + x_5 + x_7 + 2(x_8 + x_{F1} + x_{F2}) & = & N_3 \\
 x_3 + x_6 + x_7 + x_{F1} + x_{F2} & = & N_4
 \end{array}$$

# Reduced Macroscopic Model

$$\dot{\mathbf{x}} = \mathbf{MKy}(\mathbf{x})$$

$$\begin{aligned}x_3 - x_4 &= N_1 \\x_1 + x_5 + x_7 + x_8 + x_{F1} + x_{F2} &= N_2 \\x_2 + x_5 + x_7 + 2(x_8 + x_{F1} + x_{F2}) &= N_3 \\x_3 + x_6 + x_7 + x_{F1} + x_{F2} &= N_4\end{aligned}$$

The system has a **unique, positive, globally asymptotically stable** equilibrium.

*Proof:* Reaction network is **deficiency zero** and **weakly reversible**, does not admit equilibria with some  $x_i = 0$

→ We can design  $\mathbf{K}$  such that the system always converges to a target equilibrium,  $\mathbf{x}^d > \mathbf{0}$